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5-Aryl-2-furanacetic Acids

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A series of 5-aryl-2-furanacetic acids (Table I), active as antiinflammatory agents as measured by the anti-uv crythema test, have been prepared by the route outlined in Scheme I.

Table I 5-Aryl-2-furanacetic Acids

	Rel	M_{D}	Re- ccystn sol-			
Z_{i}	$\operatorname{act}.''$	°C	verit	r^d	Fortaula	Analyses
11	0.9	126 – 128	В	6.15	$C_{12}H_{10}O_3$	C, II
4-Cl	1.7	147.5 - 149	$-\mathbf{A}$	6.22	$\mathrm{C}_{12}\mathrm{H}_{9}\mathrm{ClO}_{3}$	C, H, Cl
3-Cl	0.4	109-110	В	6.20	$\mathrm{C}_{12}\mathrm{H}_9\mathrm{ClO}_3$	C, H, Cl
2-Cl	0.1	98 – 99.5	\mathbf{A}	6.20	$\mathrm{C}_{12}\mathrm{H}_{9}\mathrm{ClO}_{3}$	C, H, Cl
$4-\mathrm{Br}$	0.4	162 - 164	A	6.22	$\mathrm{C}_{12}\mathrm{H}_{9}\mathrm{Br}\mathrm{O}_{3}$	C, H, Br
4-I ⁸	0.2	116 - 117.5	\mathbf{A}	6.23	$\mathrm{C}_{12}\mathrm{H}_{9}\mathrm{FO}_{3}$	С, Н, Г
4-CH_3	0.4	143-144	В	6.23	$C_{13}H_{12}O_3$	C, H
$4\text{-CH}_3\mathrm{O}$	0.9	143 - 145.5	В	6.25	$\mathrm{C}_{13}\mathrm{H}_{42}\mathrm{O}_4$	C, H

^a Preliminary estimates; phenylbutazone = 1. ^b Melting points were taken on a Thomas-Hoover melting point apparatus in open capillaries and are uncorrected. ^c A = C_6H_{6} , B = C_6H_{6} -hexane. ^d The τ values are for the -CH₂- grouping and were determined on a Varian A-60 in CDCl₃. ^c Analyses for the elements indicated were within ±0.3% of the theoretical values.

SCHEME I

Experimental Sec ion

5-Aryl-2-furfurals, $^{\circ}$ —A mixture of 0.5 mole of the arylamine in H₂O (50 ml) and 135 ml of concentrated HCl was diazotized by the dropwise addition of 36.2 g (0.525 mole) of NaNO₂ in 100 ml of H₂O keeping the temperature below 10° by the addition of ice. After stirring at 10° for 10 min, the solution was filtered

and added all at once to a solution of 61.5 g (0.64 mole) of furfural in $\rm H_2O$ (200 ml), followed by 23 g of $\rm CnCl_2\cdot 2H_2O$ in $\rm H_2O$ (100 ml). The mixture was kept at 50–65° for 4 hr, then left standing at room temperature overnight. Volatiles were steam distilled and the black residue was taken up in either and washed (twice with 5% NaOH, $\rm H_2O$ mutil neutral). Drying (Na₂SO₄), treatment with charcoal, and removal of the solvent under reduced pressure gave the crude product which could be partially purified by crystallization from EtOH, or by distillation for those compounds which were oils. Yields were in the range of $10\text{-}55^{\prime}_{10}$.

5-Aryl-2-hydroxymethylfurans,—Reduction of the 5-aryl-2-furfurals with LiAlH₄ in 1:1 Et₂O-THF gave the crude products which were converted to the bromo derivatives without further purification.

5-Aryl-2-bromomethylfurans. A solution of 0.0282 mole of the 5-aryl-2-hydroxymethylfuran in 65 ml of Et₂O was cooled in an ice bath. To this was added dropwise a solution of 2.8 g (0.0103 mole) of PBr₈ in Et₂O (20 ml). After the addition was complete, the mixture was allowed to stir at room temperature for 1 hr. The ether was then decanted and the gunday residue was washed (Et₂O). The combined ether extracts were swirled with cold 50% NaOH, decanted, and dried (solid KOH). The solvent was removed under reduced pressure at room temperature. The mustable nature of the bromomethyl compounds necessitate their immediate conversion to the nitriles.

5-Aryl-2-cyanomethylfurans. The crude 5-aryl-2-bromomethylfuran from 0.0282 mole of the hydroxymethyl compound was dissolved in 50 nd of acctone and treated with 1.5 g (0.03 mole) of NaCN in 10 nd of $\rm H_2O$ and the solution was heated at reflux for 3 hr. Work-up of the dark reaction mixture in the usual manner gave the crude nitrile as a dark, viscous oil.

5-Aryl-2-furanacetic Acids,—The crude nitrile (5 g) in EtOH (100 ml) was treated with 5 g of KOH in 25 ml of H₂O and the resulting solution was heated at reflux for 6 hr. Work-up in the usual manner gave the crude acid as an oil which was chromatographed on silica gel. After elution of some colored material with benzene, the product was eluted with 10°C ether in benzene. Recrystallization gave the pure 5-aryl-2-furanacetic acids.

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The Preparation and Pharmacology of Some 11β -Hydroxy-4-methylestratrienes

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Recently we reported that 17α -ethynyl-1,4-dimethylestra-1,3,5(10)-trien-17 β -ol (IIIa) and its acetate IIIb showed antiinflammatory properties in the carrageenin-induced foot edema rat assay and that both of these substances also reduced the plasma cholesterol concentration of rats made hypercholesterolemic with propylthiouracil.¹ Earlier, Goldkamp, et al., observed that estra-1,3,5(10)-trien-17-ones—and— 17α -ethynylestra-1,3,5(10)-trien-17 β -ols with a methyl group attached to ring A had a favorable lipodiatic—estrogenic ratio.² These findings prompted us to determine whether estratriene derivatives with an oxygen function at C-11, but not in ring A. also possess antiinflammatory and antiatherogenic effects.

11-Oxygenated corticosteroids are systemically active

C. V. Winder, J. Wax, V. Burr, M. Been, and C. E. Rosiere, Arch. Int. Pharmacodyn. Ther. 116, 261 (1958).

^{(2) (}a) R. Oda, Mem. Fac. Eng. Kyoto Univ., 14, 195 (1952); Chem. Abstr., 48, 1935 (1954); (b) H. Akasld and R. Oda, Rept. Inst. Chem. Res. Kyoto Univ., 19, 93 (1949); Chem. Abstr., 45, 7519 (1951); (c) C. S. Davis and G. S. Lougheed, J. Heterocycl. Chem., 4, 153 (1967).

⁽¹⁾ L. J. Chiou, J. Med. Chem., 9, 602 (1966).

⁽²⁾ A. H. Goldkamp, W. M. Goelm, R. A. Mikulee, E. F. Nutting, and D. L. Cook, ibid., 8, 469 (1965).